

A04-1 Theory and Computation

Member

Ab initio

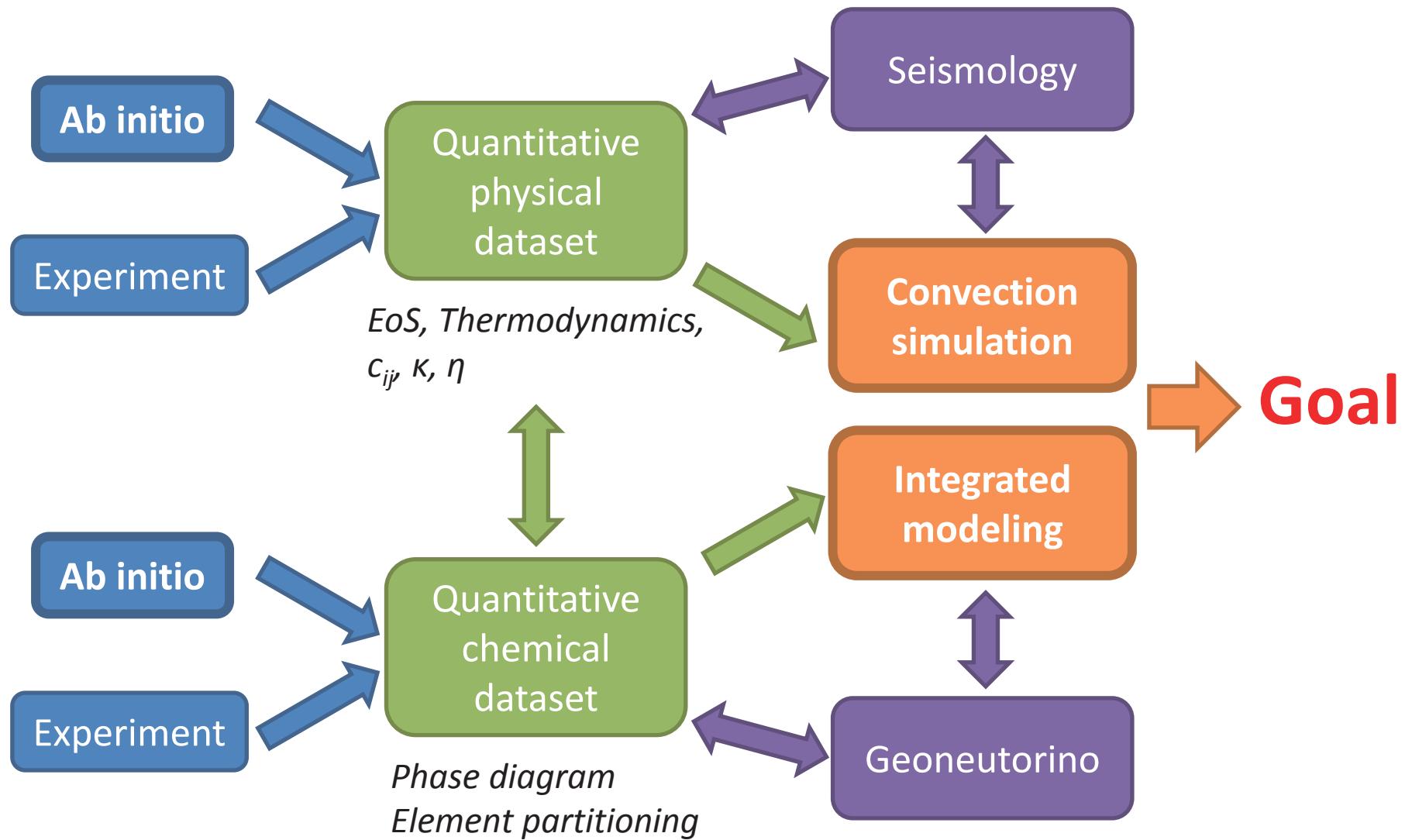
- T. Tsuchiya (Ehime Univ, TiTech) Element partitioning & Diffusion
- J. Tsuchiya (Ehime Univ, TiTech) Hydrous phase
- H. Dekura (Ehime Univ) Thermal transport

Hydrodynamics and others

- F. Takahashi (Kyushu Univ) Core dynamics
- T. Miyagoshi (JAMSTEC) Mantle and Core dynamics
- T. Nakagawa (JAMSTEC) Core-mantle coupling
- J. Hernlund (TiTech) Integrated model

+ Kobo Researchers

Research structure



Key properties

Equilibrium

1. Structure search
2. Equation of state
3. Phase equilibrium
4. Thermodynamics
5. Elastic tensor
6. Element partitioning

Transport

1. Thermal conduction
2. Electrical conduction
3. Atomic diffusion

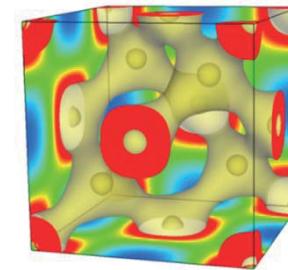
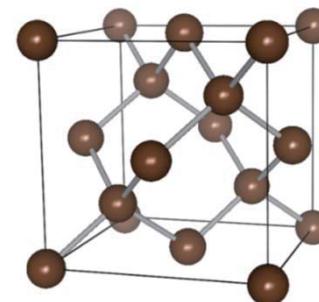
Ab initio mineral physics

Density Functional Theory
(Hohenberg & Kohn 64; Kohn & Sham 65)

$$\left[-\frac{\hbar^2}{2m} \Delta - \sum_I \frac{Z_I}{|\mathbf{r} - \mathbf{R}_I|} + e^2 \int \frac{n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r}' + V_{XC}[n(\mathbf{r})] \right] \varphi_i(\mathbf{r}) = \epsilon_i \varphi_i(\mathbf{r})$$

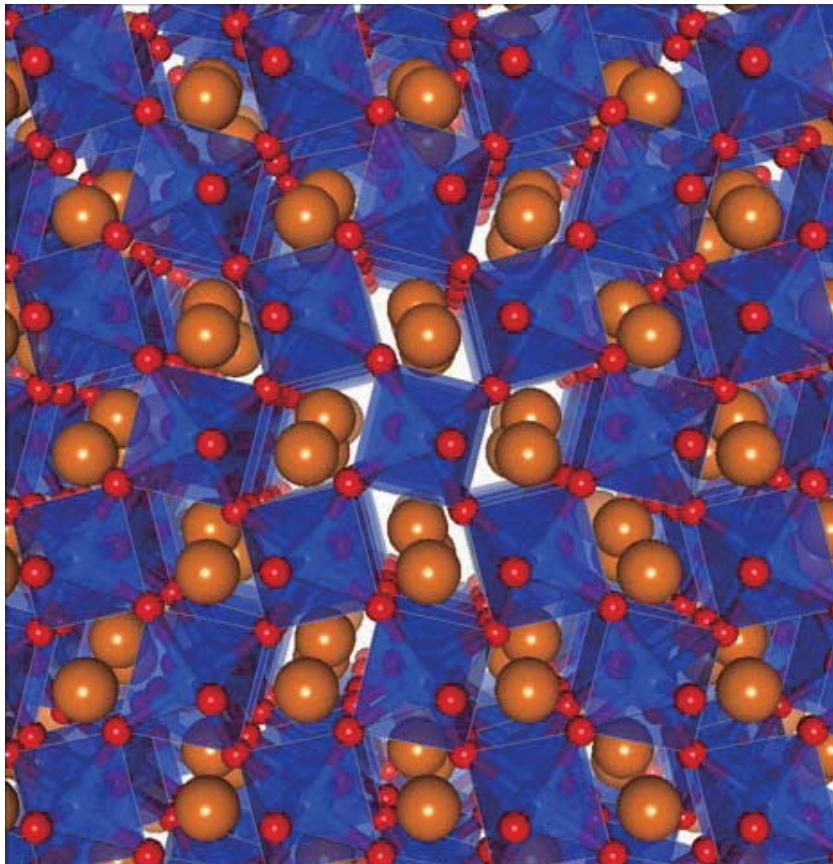
$$n(\mathbf{r}) = \sum_i |\varphi_i(\mathbf{r})|^2$$

φ : 1e wavefunction
 n : e density

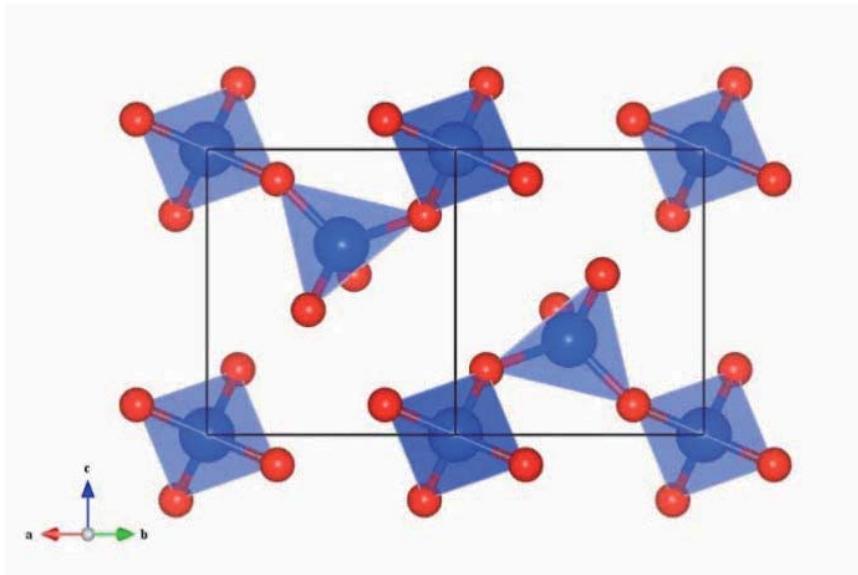


Structure search

MgSiO_3 Bridgmanite



SiO_2 α -Quartz

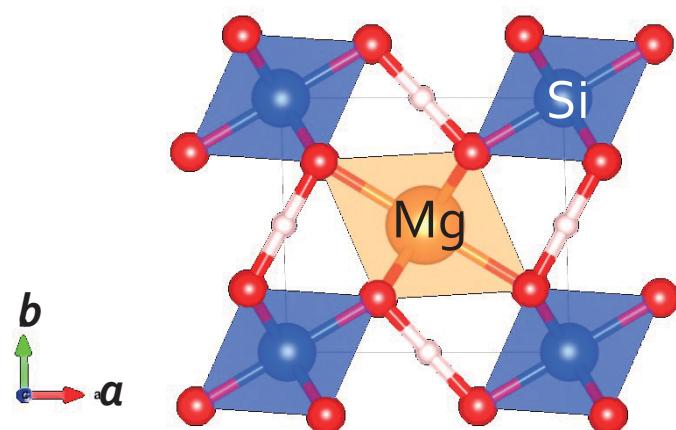


- Mg
- Si
- O

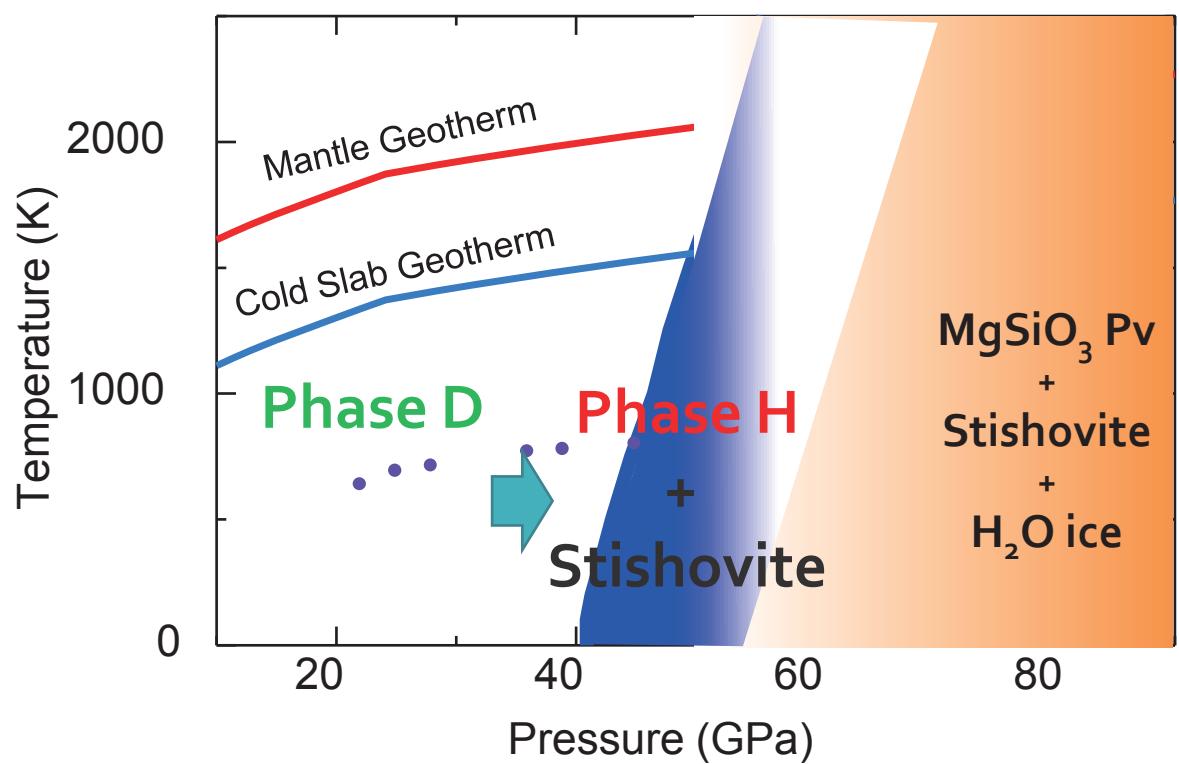
*Tsuchiya+ (04) EPSL
Tsuchiya & Tsuchiya (11) PNAS*

New high-pressure hydrous phase H

Phase D ($\text{MgSi}_2\text{O}_6\text{H}_2$) → phase H (MgSiO_4H_2) + SiO_2

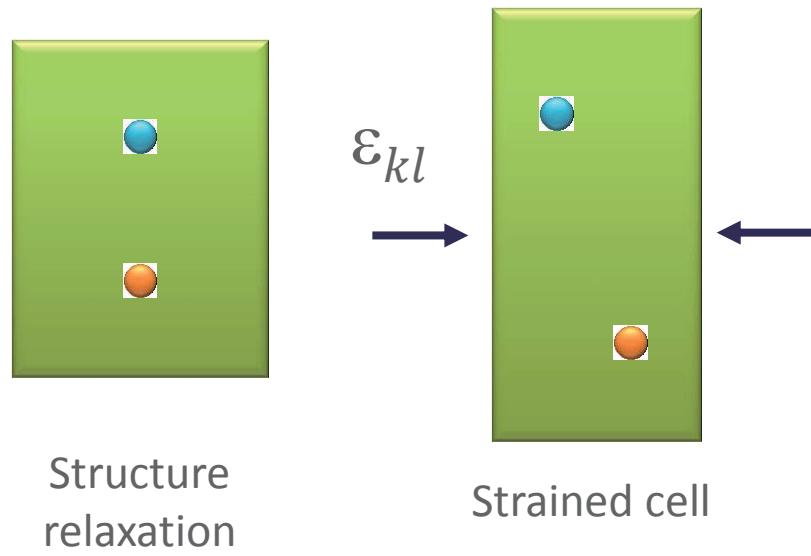


~ 45 GPa



J. Tsuchiya (13) GRL
Nishi+ (14) Nature Geosci.

High- P, T elasticity



$$c_{ij}^S(P, T) \rightarrow K_S \text{ & } \mu \text{ (Hill average)}$$

$$V_P = \sqrt{\frac{K_S + 4/3 \mu}{\rho}}, V_S = \sqrt{\frac{\mu}{\rho}}, V_\Phi = \sqrt{\frac{K_S}{\rho}}$$

Method-1: Stress-strain Molecular Dynamics

$$\sigma_{ij}(P, T) = c_{ijkl}(P, T)\varepsilon_{kl}$$

Method-2: Energy-strain Lattice dynamics + Quasiharmonic approx

$$c_{ij}^T(P, T) = \frac{1}{V} \left(\frac{\partial^2 G(P, T)}{\partial \varepsilon_i \partial \varepsilon_j} \right)_T$$

Isothermal – adiabatic conversion

$$c_{ij}^S(P, T) = c_{ij}^T(P, T) + \frac{VT\lambda_i\lambda_j}{C_V}$$

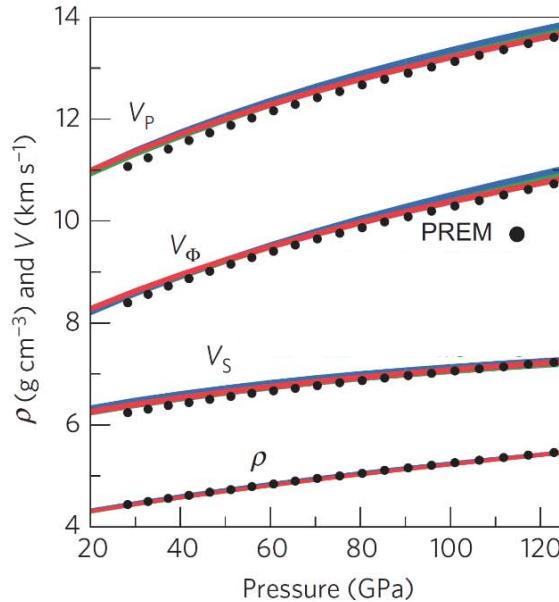
$$\lambda_i(P, T) = \left(\frac{\partial S(P, T)}{\partial \varepsilon_i} \right)_T$$

Elastic wave velocity of aggregates



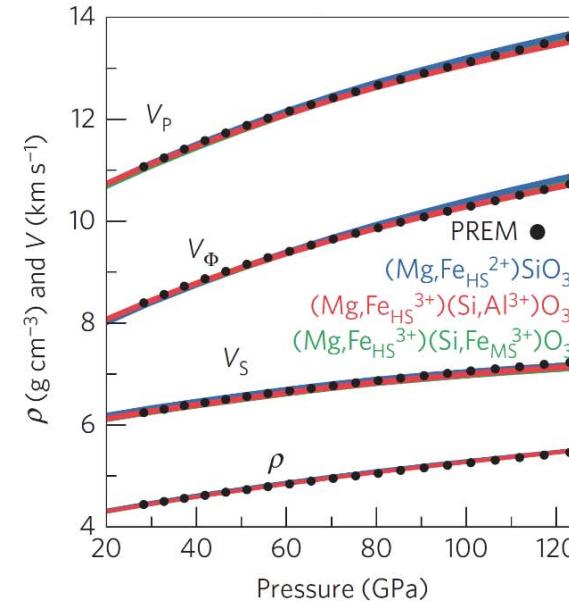
Br:Fp = 9:1

Chondrite



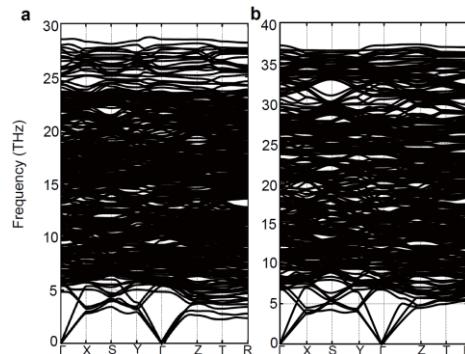
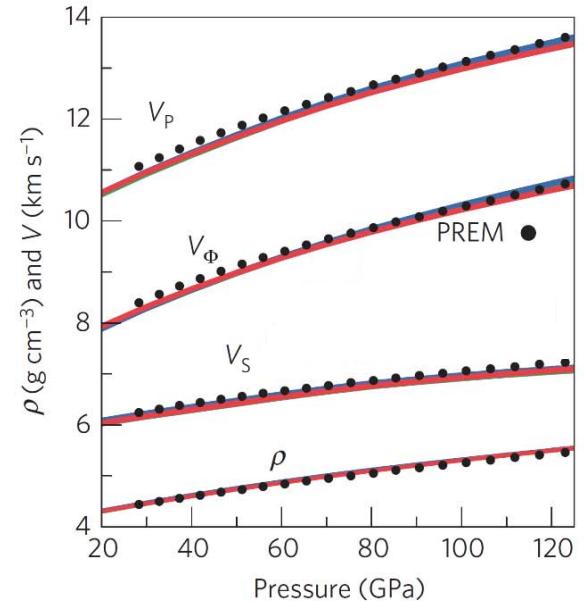
8:2

Pyrolite



7:3 in vol

Olivine



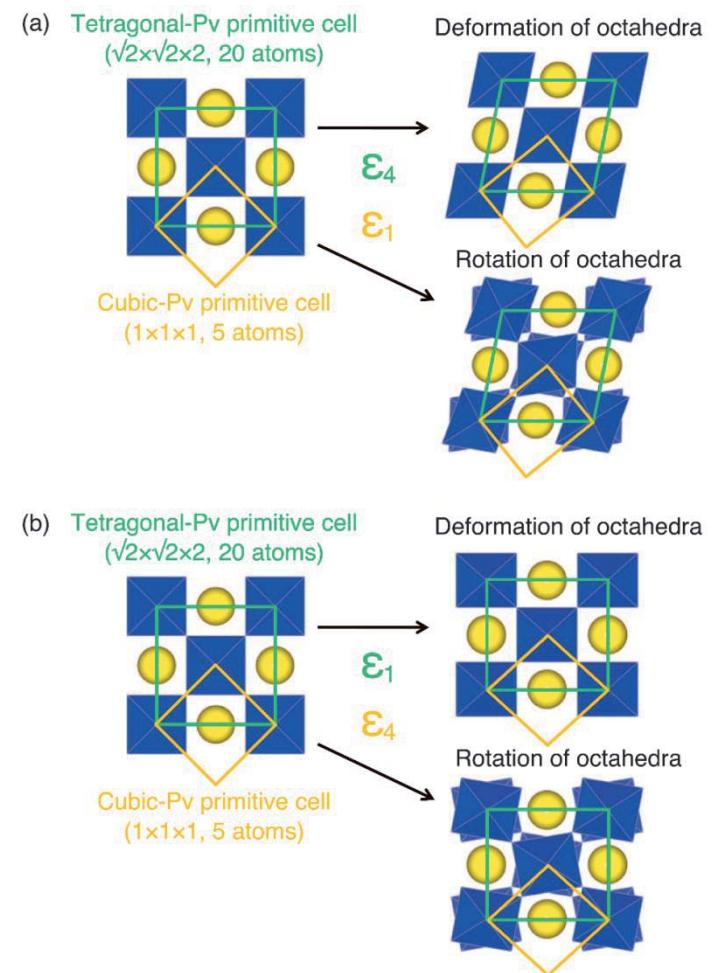
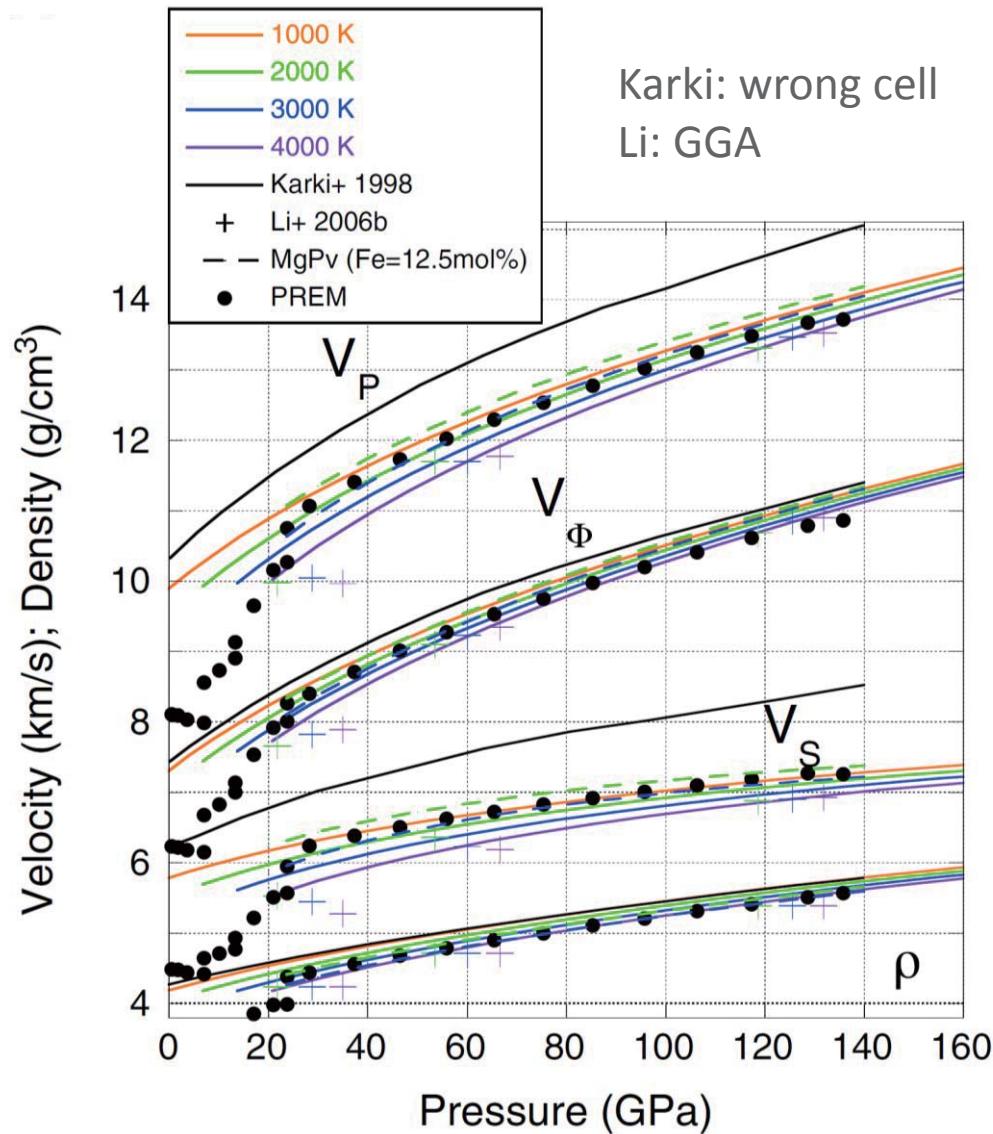
Internally consistent LSDA+U

+

Direct LD

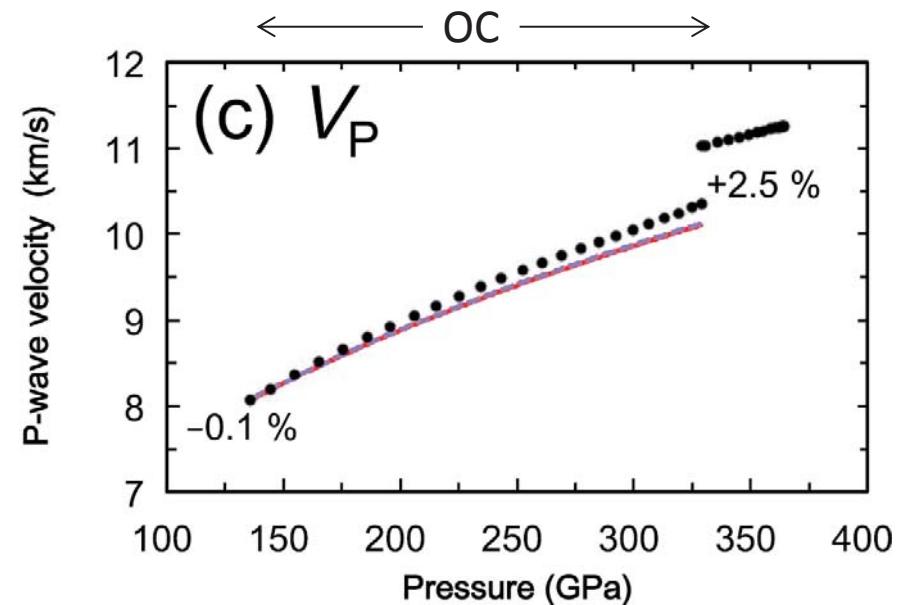
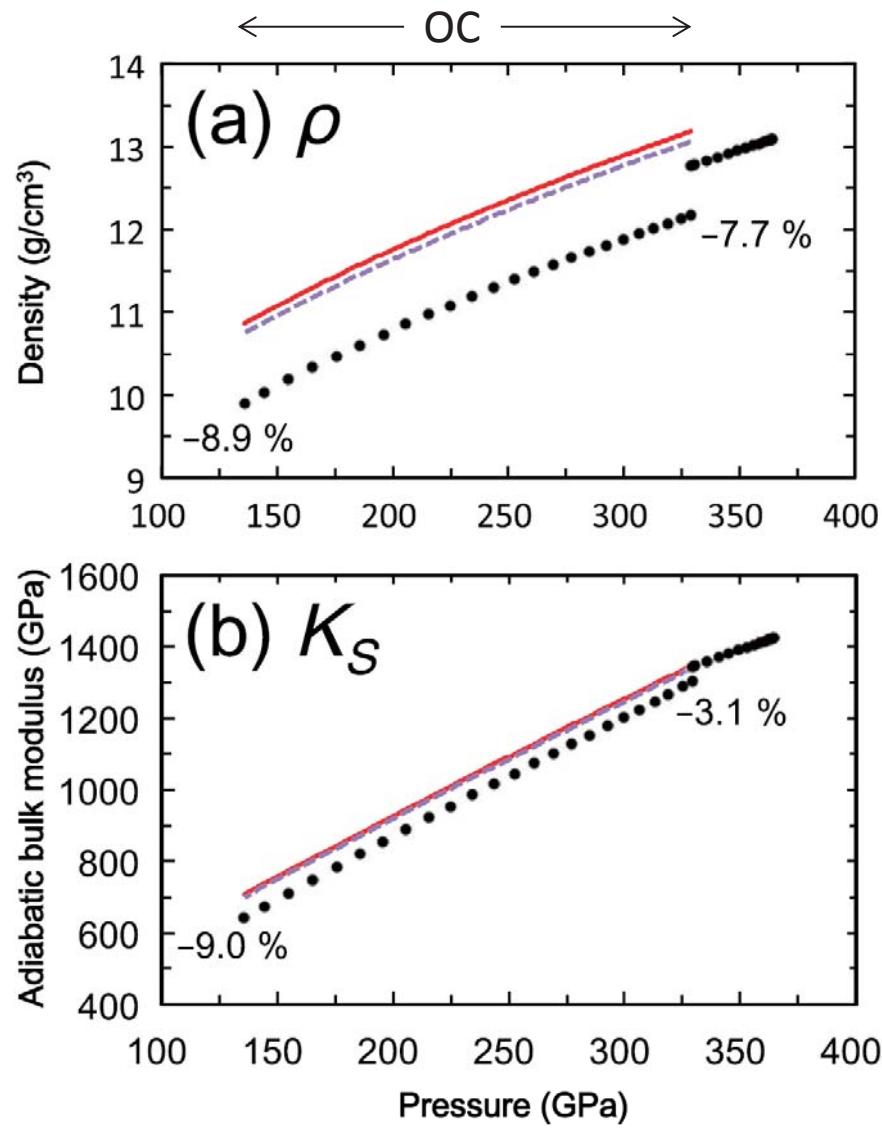
Wang, Tsuchiya & Hase (15) Nature Geosci.

CaSiO₃ Pv



Kawai & Tsuchiya (2014) JGR
Kawai & Tsuchiya (2015) GRL

Liquid Fe



Liquid Fe alloy

Model:

Fe + O, S, Si, C, H, Ni

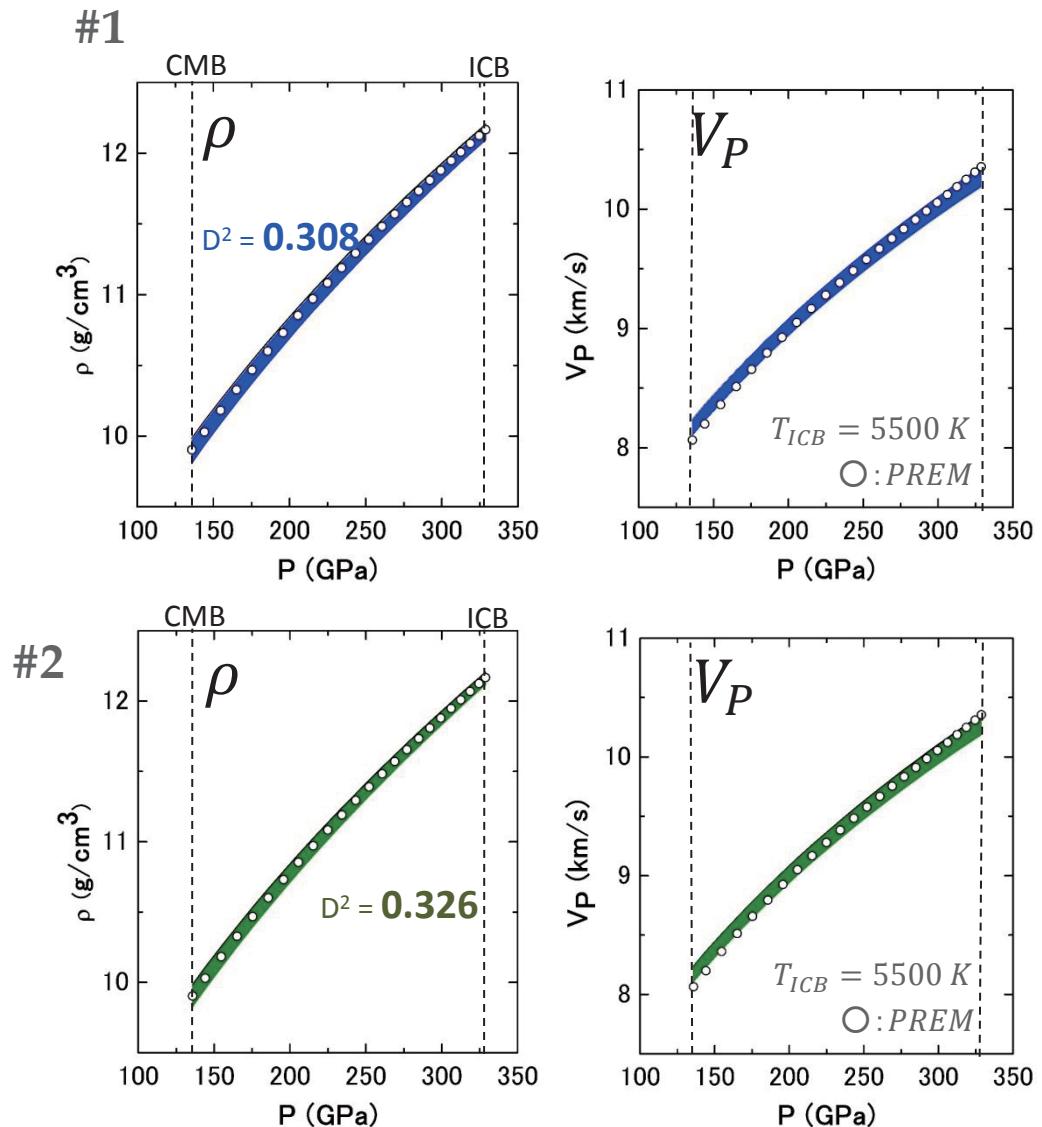
$T_{ICB} = 5000, 5500, 6000$

Best compositions

#1: $\text{Fe}_{0.71}\text{Ni}_{0.05}\text{Si}_{0.03}\text{O}_{0.21}$

#2: $\text{Fe}_{0.71}\text{Ni}_{0.05}\text{S}_{0.04}\text{O}_{0.20}$

Ohsumi+, in prep.



Lattice thermal conductivity

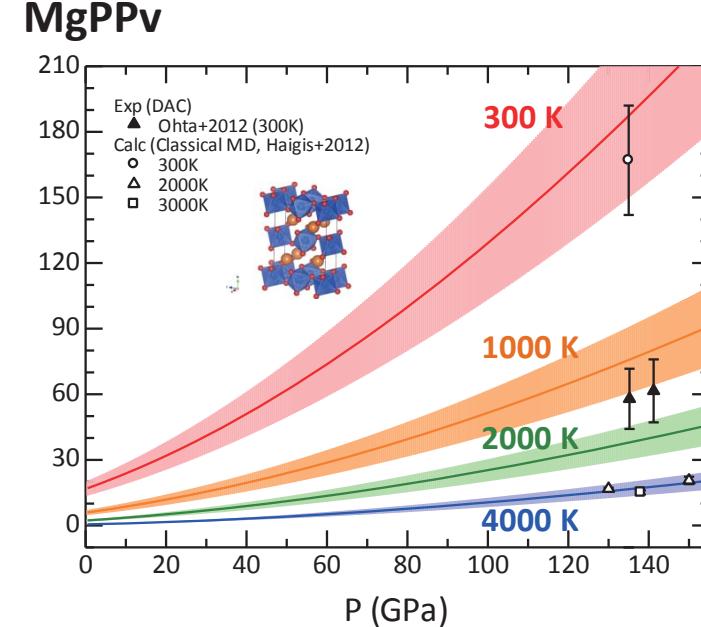
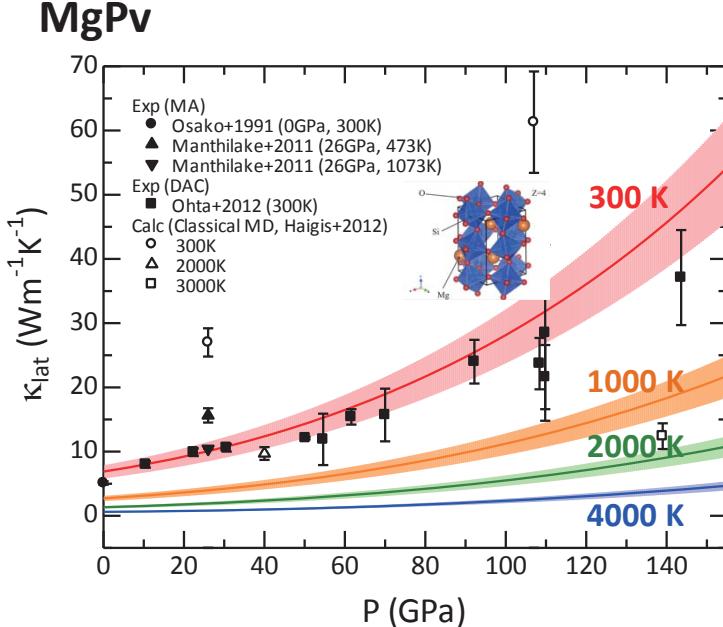
$$\kappa_{\text{lat}} = \frac{1}{3} \sum_s^{3n} \int \mathbf{v}_{\mathbf{q},s}^2 c_{\mathbf{q},s} \tau_{\mathbf{q},s} d\mathbf{q}$$

$$\tau_{\mathbf{q},s} = \frac{1}{2\Gamma(\omega_{\mathbf{q},s})}$$

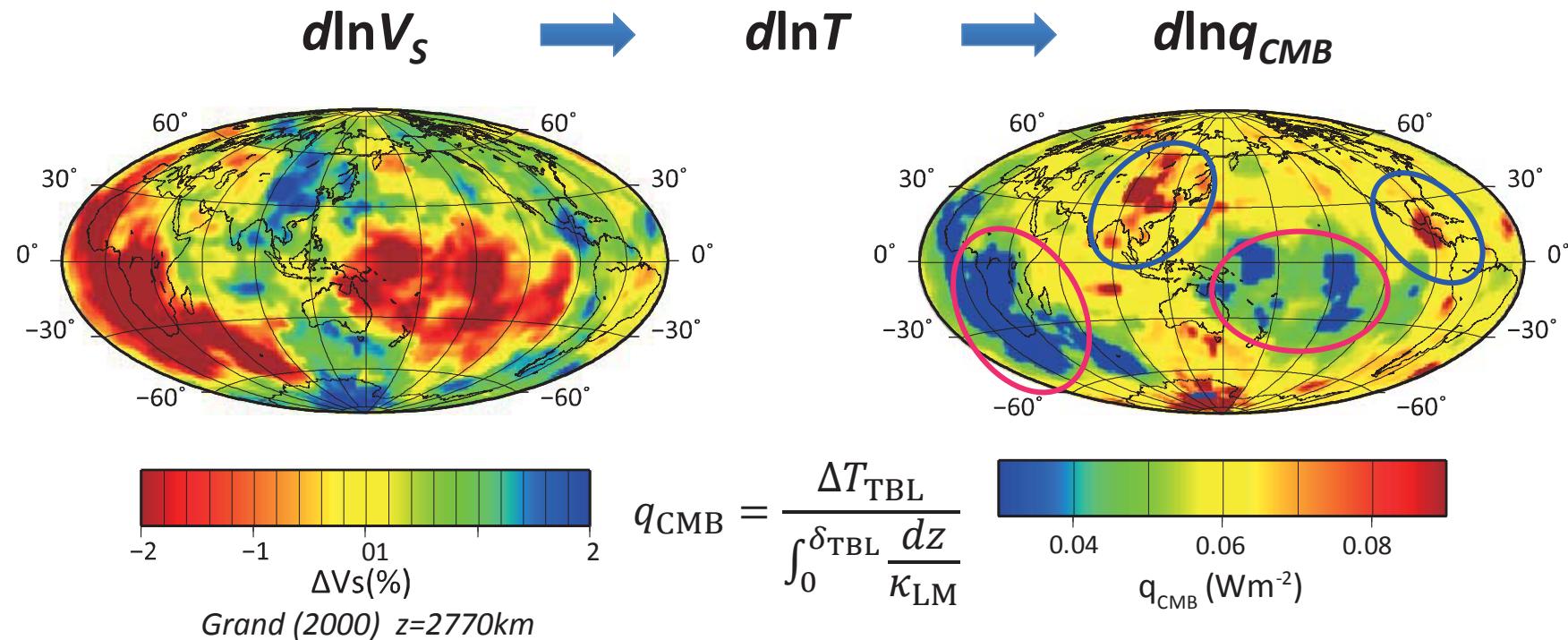
Third-order dynamical tensor

$$\begin{aligned} \Gamma_{\mathbf{q},s}(\omega) = & \frac{\pi}{2N_{\mathbf{q}'}} \sum_{\mathbf{q}',s',s''} |V_3(\mathbf{q}_s, \mathbf{q}'s', \mathbf{q}''s'')|^2 \\ & \times [\{1 + n_{\mathbf{q}',s'} + n_{\mathbf{q}'',s''}\} \delta(\omega_{\mathbf{q}',s'} + \omega_{\mathbf{q}'',s''} - \omega) \\ & + 2\{n_{\mathbf{q}'',s''} - n_{\mathbf{q}',s'}\} \delta(\omega_{\mathbf{q}',s'} - \omega_{\mathbf{q}'',s''} - \omega)] \end{aligned}$$

$$\mathbf{q}'' = -\mathbf{q} - \mathbf{q}' + \mathbf{G}$$



Local CMB heat flux interpreted from V_S tomography

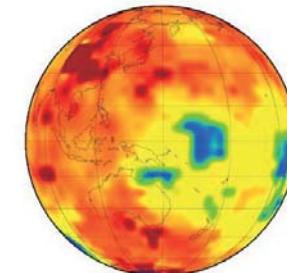


Hot regions = Low heat flux ($\sim 0.03 \text{ W/m}^2$)
 Cold regions = High heat flux ($\sim 0.09 \text{ W/m}^2$)

Total CMB heat flow estimated using κ of pure phases

$$J_{CMB} = \int dS q_{CMB}$$

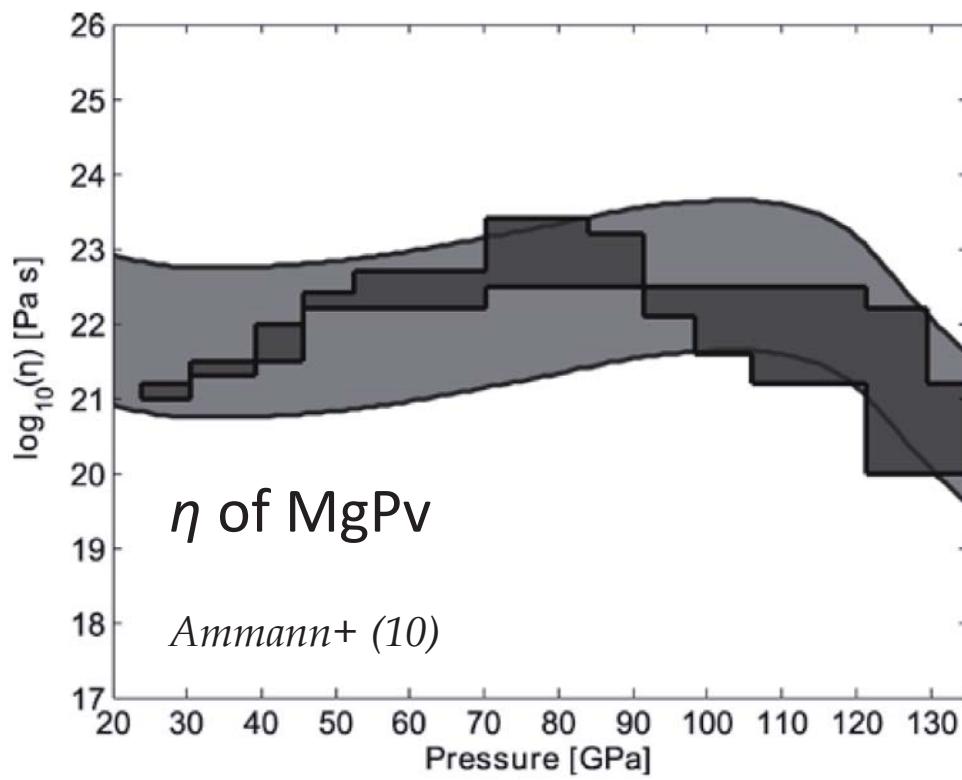
$\sim 9 \text{ TW}$



Viscosity (diffusion creep)



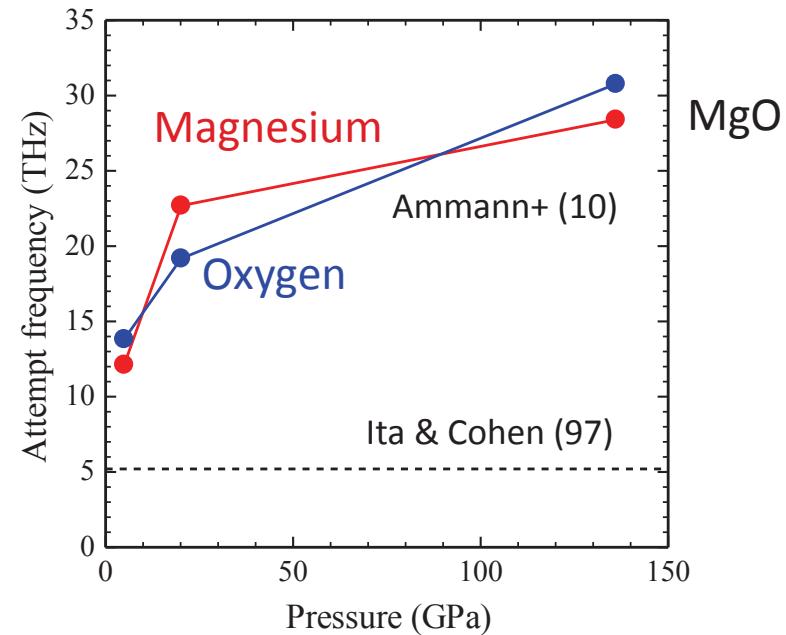
Nabarro-Herring viscosity



$$\eta \propto \frac{d^2 k_B T}{D_{eff} \Omega}$$

$$D = N_V \frac{Z_m}{6} l^2 \nu \exp\left(-\frac{\Delta H_m}{k_B T}\right)$$

↑
Trial frequency



Application to element partitioning

Partitioning condition

$$\mu_{\text{component } i}^{\text{phase } A} - \mu_{\text{component } j}^{\text{phase } A} = \mu_{\text{component } i}^{\text{phase } B} - \mu_{\text{component } j}^{\text{phase } B}$$

Chemical potential

$$\mu_i = \left(\frac{\partial G}{\partial n_i} \right)_{P,T,N}$$

Free energy

$$G(P, V, \{n_i\}) = H(P, V, \{n_i\}) - \underbrace{TS(P, V, \{n_i\})}_{\substack{\uparrow \\ \text{Entropy:} \\ \text{Hard to estimate}}}$$

Thermodynamic integration

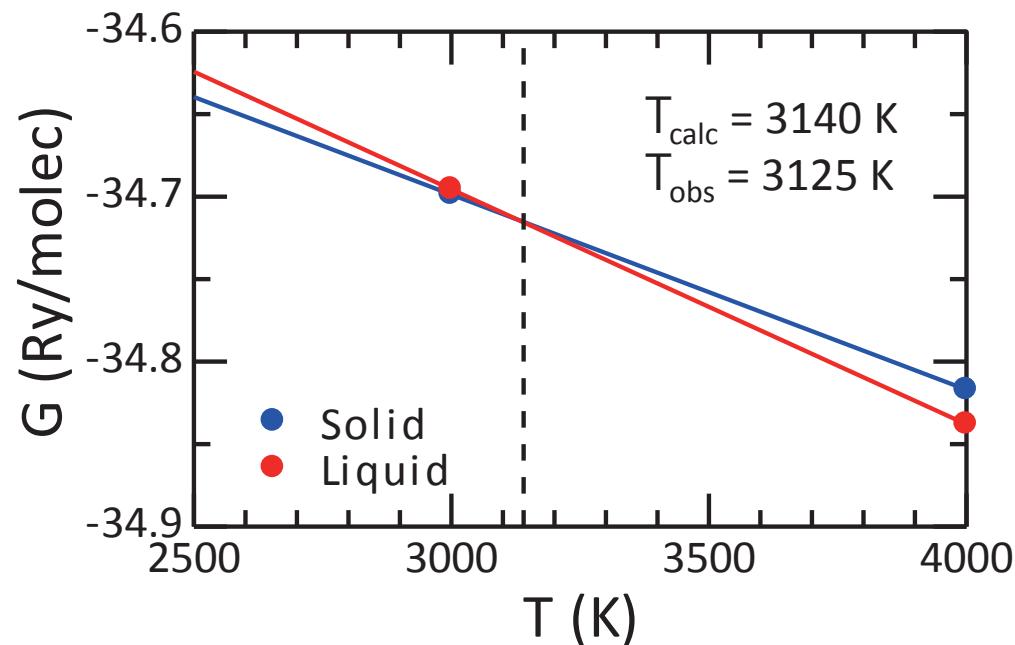


$$F_1 = F_2 + \int_0^1 \left\langle \frac{\partial U}{\partial \lambda} \right\rangle d\lambda$$

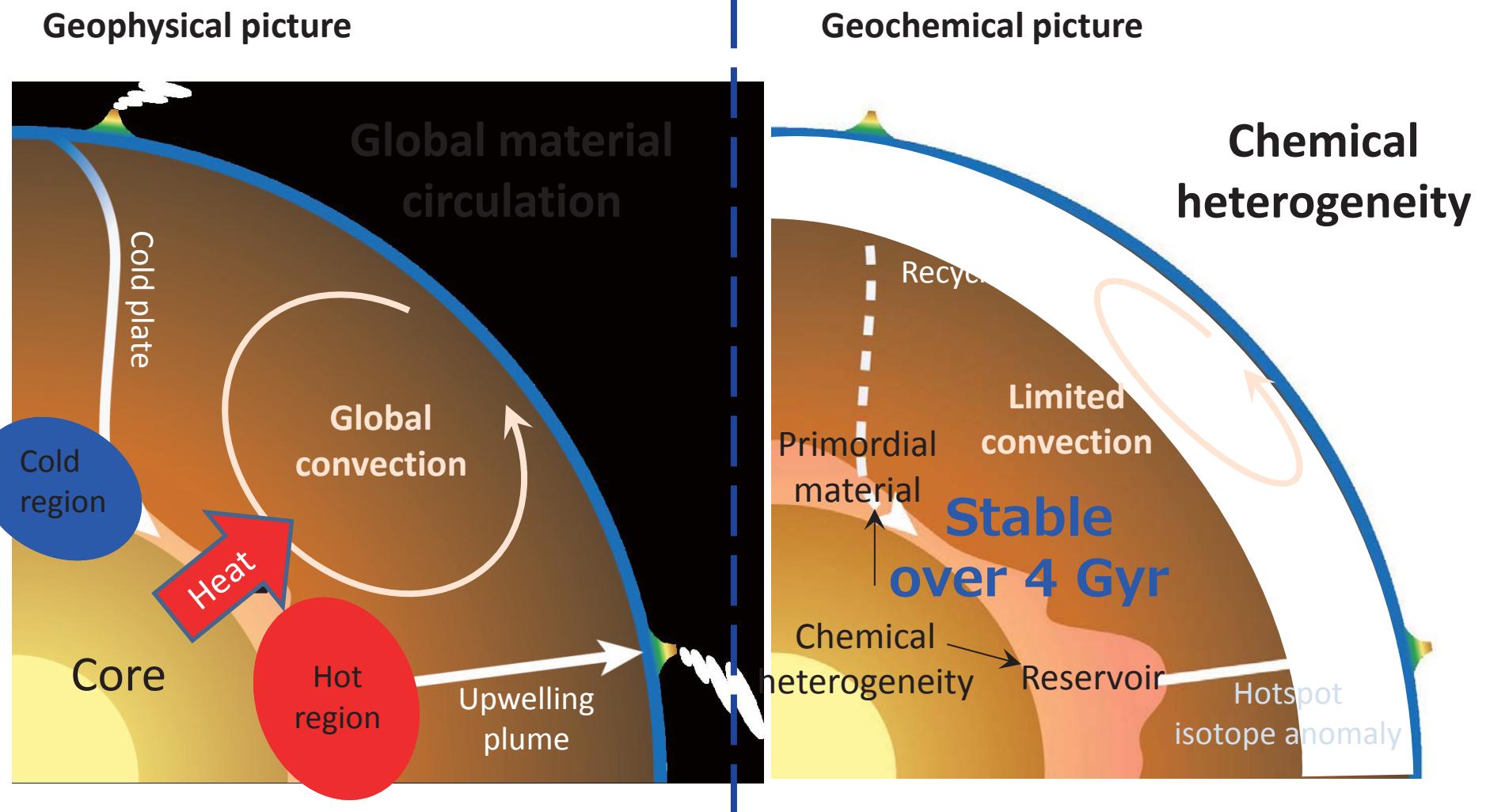
Potential energy of a combined system $U(\lambda) = \lambda U_1 + (1 - \lambda) U_2$

We have developed one-step (ideal \rightarrow ab initio) TI-MD code (Taniuchi+ in prep)

Melting point of MgO
at 0 GPa

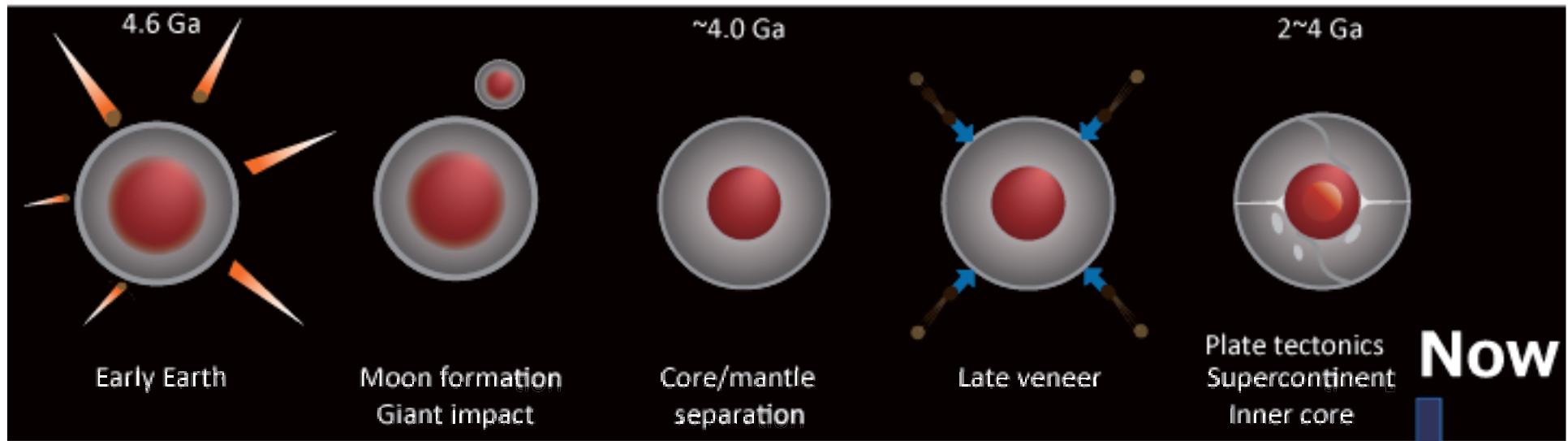


Chemical heterogeneity/Heat source distribution



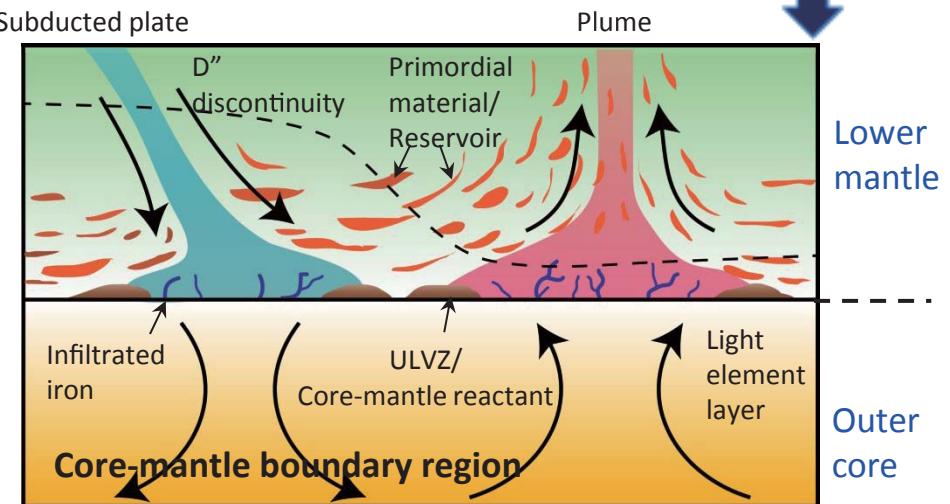
Details/ Origin/ Evolution

Formation processes of chemical heterogeneity



Time & Scale?

Quantitative data of
high-*P,T* element partitioning
and thermal property



Many factors of chemical heterogeneity